GAPHYOR Data Center http://gaphyor.lpgp.u-psud.fr

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The GAPHYOR Data Center produces a Database on the properties of atoms, molecules and neutral or ionized gases, including Chemical Reactions, with about 500,000 entries.

1 Structure of a GAPHYOR record

The structure of a GAPHYOR entry was originally conceived to describe collisions/reactions in gases. It now describes also simpler (energy levels, ...) or more complex (gas-surface interactions, clusters, ...) phenomena. Every GAPHYOR entry includes the following fields:

 $/se/M1,M2,M3/pr/M4,M5,M6/\\ /info/enva/val/rem/\\ /an/jo,vo,page/au1,au2,...au10/nastci/$

2 Initial and Final state: M1 M2 M3 - M5 M6 M7

Atomic and molecular reactants: /n form ix/s/

- form Chemical formula: Example /H1C3N1O2/.

 1 to 4 elements, no more than 8 atoms in the molecule.
- i Index of ionization, x Index of excitation, s Index of internal structure.
- n Multiplicity of the Reactant (e.g. 2 H atoms in initial state).

Electrons, photons, fields, particles.

Special Reactants:

- Iso-electronic series. Example: Aa(He)Z(02-12)
- Clusters. Examples: He(H2O)3/+, (C)60
- Surface reactants (solid/liquid walls, adsorbed atoms/molecules).

3 Sections SE and Processes PR

GAPHYOR is a "broad band" Database organized into 5 Sections:

- 1 Properties of isolated atoms and molecules
- 2 Photonic Collisions
- 3 Electronic Collisions
- 4 Collisions between atoms or molecules
- 5 Macroscopic properties of Gases

	Struct.	hv coll.	e coll.	${ m At.mol.coll.}$	Mac.prop.	Σ
Fact.Data	267178	26941	29743	124753	32006	480621
Num.Data	6642	347	1650	9010	734	18383
Σ	272730	27244	30810	132107	32642	495533

Table 1: Statistics of Data (June 15th, 1997)

The list of Processes (PR) included in every Section (SE) is given in Appendix 1.

4 Additional information INFO

- Type of Data: S Total and transport cross sections, R Reaction rate constants, ...
- Nature of Data: T Theoretical data, E Experimental data, ...
- Energy domains: L Thermal, J Medium, H High energies
- Special information: 8 Clusters, / Gas-surface interactions, ...

5 Numerical Data

- \bullet ENVA gives the energy, or energy range covered by the Data. Example: 10 50 100 eV
- VAL gives the values them-selves (cross-sections, ...). Example: 3.5 12.2 6.4 E-16 cm²
- **REM** gives additional information (e.g. spectroscopic notations).

Note that the general idea is to give *simple information* (eg. lifetimes, reaction rates), or some orders of magnitude.

6 Internet server

GAPHYOR WWW server is now well developed and user-friendly. Its main features are:

- Query forms. Two query forms (simple and advanced search) are offered.
- **Intermediate report**. This gives a sketch of the first results of a search and offers some specific additional choices to the user.

• Data displays. Two displays are offered: a full display of all the found entries (possibly selecting the most recent ones), or, for large Data sets, a tabular display obtained by downloading a postscript file, which can then be printed locally.

Simplified statistics on User Locations are shown below.

na	data sets	org.na	data sets
fr	6098	u-psud.fr	1099
??	2293	uni-heidelberg.de	857
de	1194	${ m cnrs-imn.fr}$	597
ru	1151	univ-lille1.fr	580
us	993	${ m cea.fr}$	535
jр	530	onera.fr	337
	•••	•••	•••
Σ	14 885		14 885

Table 2: Nations and Organizations using GAPHYOR

7 GAPHYOR Handbook

GAPHYOR is going to publish a new (fully redesigned) version of the HANDBOOK. The Volumes will be grouped in two Series:

- Chemical Volumes (17) describing the properties of a given group of Chemical systems.
- Thematic Volumes (9) describing a given group of Processes for all Chemical Systems.

The publication timing will be a function of the various orders from users for one or several specific Volumes. The sizes are small, so as to make it easy for every user to order only the Volumes covering their own field of interest.

The orders (see form in the Web) will be satisfied within a period of one month after receipt of payment.

	Chemical Volumes	Selection	Pages	Price \$
C01	Hydrogen, Deuterium, Tritium	62%	130	12
C02	H + Noble Gases (He, Ne, Ar, Kr, Xe, Rn) +	72%	130	12
C03	H + Mono-, bi- and trivalent elements $+$	60%	130	12
C04	H + Carbon/Titanium Groups +	48%	410	30
	see List in the Web	•••	•••	•••
C17	Surfaces/Clusters	52%	130	12
	Thematic Volumes			
T00	Cross Sections/Reaction Rates (Numerical values)	100%	110	10
	see List in the Web	•••		
T30	Electronic Collisions (Section 3)	62%	210	18
T50	Macro.Prop. of Gases (Section 5: Transport coefficients,)	58%	210	15
		Σ	4640	321

Table 3: Handbook Volumes

8 Appendix 1

Section 1. Structure

EN	Energy	levels		wave	functions
	LIICI 5.9	ICVCID	9	wave	Idiiculons

EA Unstable energy levels

EI Energy of isomerization

VR Potentiel curves, structure of molecules

DP Dipolar moments

PE Electric (or magnetic) polarizability

TR Radiative transition (probability,...)

IN Autoionization

DT Autodetachment

Section 2. Photonic collisions

SN Effective absorption, total diffusion

AN True absorption

SC Angular diffusion (scattering)

EL Elastic diffusion (Thomson, Rayleigh)

DO Depolarization, Change of polarization

EX Photoexcitation

ER Emission of line

DX Photodeexcitation (stimulated emission)

FF Free-free absorption(inv. bremsstrahlung)

IN Photoionization

IM Creation of an ion pair (positive-negative)

 ${\bf DT\ Photodetachment}$

DS Photodissociation

EE Photoemission of electrons (ions) by solids

Section 3. Electronic Collisions

SN Total cross sections

SM Transport cross sections (momentum, ...)

EL Elastic collision

DO Depolarization, Change of polarization

EX Excitation

DX Deexcitation

BS Bremsstrahlung

IN Ionization

IM Creation of an ion pair (positive-negative)

DT Detachment

AT Attachment

RR Radiative recombination

RD Dielectronic recombination

RE e-e-i recombination

RO e-i-o recombination

RS Dissociative recombination

DS Dissociation

DG Desorption

EE Emission of electrons by a solid

PU Emission of neutrals or ions by solids (sputtering)

Section 4. Atom./mol. Coll. (and Reactions)

EN Energy or enthalpy of reaction

KE Constant of chemical equilibrium

SN Total cross sections

SM Transport cross sections (momentum, ...)

SC Angular diffusion (scattering)

SP Stopping power

EL Elastic collision

DO Depolarization, change of polarization

EX Excitation

DX Deexcitation (quenching)

TE Excitation transfer

IN Ionization

IM Creation of an ion pair (positive-negative)

DT Detachment

TI Ionizing charge transfer

RI Recombination ion-ion

CX Charge transfer

XD Dissociative charge transfer

CA Capture of electrons

SR Loss of electrons (stripping)

DS Dissociation

IR Interchange reaction (of one or several atoms)

IA Associative interchange reaction

ID Dissociative interchange reaction

AS Association

DG Desorption

AD Adsorption

EE Emission of electrons by a solid

PU Emission of neutrals or ions by solids (sputtering)

Section 5. Macroscopic properties

ST Statistics of levels

FT Thermodynamic functions

PV Compressibility, equation of state

PE Dielectric and magnetic constants

DN Diffusion

VI Viscosity

CT Thermal conductivity

LW Line broadening and shift (collisional effects)

PI First coefficient of Townsend

DT Detachment

AT Attachment

FE Distribution function of electrons

ME Mobility of electrons

CE Electrical conductivity

DE Diffusion of electrons

MD Characteristic temperature of electrons (D/μ)

PC Power delivered by electron-neutral collisions

FI Distribution function of ions

MI Mobility of ions

DI Diffusion of ions

DA Ambipolar diffusion

Remark This list is slightly simplified (see the Web)